

**SABA BANANA (*Musa acuminata*) PEEL AS BIOSORBENT OF CHROMIUM  
(Cr) HEAVY METAL FROM AQUEOUS SOLUTION**

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# TABLE OF CONTENTS

	PAGE
	i
TITLE PAGE	ii
ACCEPTANCE SHEET	iii
BIOGRAPHICAL SKETCH	iv
ACKNOWLEDGEMENT	vii
TABLE OF CONTENTS	ix
LIST OF TABLES	x
LIST OF FIGURES	xii
LIST OF APPENDIX TABLES	xii
LIST OF APPENDIX FIGURES	xiii
ABSTRACT	1
INTRODUCTION	1
Background of the Study	1
Statement of the Problem	4
Objectives of the Study	5
Significance of the Study	6
Scope and Limitation of the Study	7
Time and Place of the Study	7
REVIEW OF RELATED LITERATURE	8
Saba Banana ( <i>Musa acuminata</i> )	8
Metal Contaminants	9
Chromium	10
Different Treatment in Adsorbing Heavy Metals	11
Removal of Heavy Metals through Adsorbents	12
METHODOLOGY	14
Conceptualization of the Study	14
Instrumentations	17

Scanning Electron Microscopy (SEM)	17
Materials and Methods	17
Collection and preparation of materials	18
Recovery of material (R)	18
Preparation of Chromium solution	19
Characterization of biosorbent	19
Adsorption Experiment	19
Amount of Chromium ion adsorbed (A)	20
Ion uptake capacity	20
Percentage of removed Chromium (P)	22
RESULTS AND DISCUSSION	22
Collection and preparation of materials	22
Recovery of material (R)	23
Preparation of Chromium solution	23
Characterization of the SBP biosorbent	25
Surface Morphology Analysis of the SBP biosorbent	26
Adsorption Experiment	29
Amount of Ions Adsorbed, A (mg/L)	29
Ion uptake capacity, q (mg/g)	32
Percentage of removed Chromium, P (%)	34
SUMMARY, CONCLUSION AND RECOMMENDATION	37
Summary	37
Conclusion	39
Recommendations	40
LITERATURE CITED	42
APPENDICES	47

## LIST OF TABLES

TABLE		PAGE
1	Recovery of material	23
2	Amount of Chromium ions adsorbed by SBP biosorbent	29
3	Ion uptake capacity of SBP Biosorbent	32
4	Percentage of removed Chromium (VI)	34

## LIST OF FIGURES

FIGURE		PAGE
1	Conceptual framework of the study	15
2	Flowchart of activities	16
3	Buying of ripe saba banana in the local market	22
4	Preparation of Chromium solution	24
5	Different mass of Potassium dichromate	24
6	Chromium solution prior to adsorption experiment	25
7	SU3800 Scanning Electron Microscope at CLSU Nanotechnology Research and Development Facility	26
8	Micrograph of dried SBP biosorbent at 650x magnification showing the processed image using SEM- EDX	27
9	Micrograph of dried SBP biosorbent at 1500x magnification showing the processed image using SEM- EDX (1)	27
10	Micrograph of dried SBP biosorbent at 650x magnification showing the processed image using SEM- EDX (2)	28
11	Amount of Chromium ions adsorbed by SBP biosorbent	31
12	Ion uptake capacity of SBP biosorbent	33
13	Percentage of removed Chromium (VI) using dried SBP biosorbent	35

## LIST OF APPENDIX TABLES

APPENDIX		PAGE
1	Amount of ions adsorbed by SBP biosorbent	48
2	Ion uptake capacity of SBP biosorbent	48
3	Percentage of removed Chromium (IV)	48

## LIST OF APPENDIX FIGURES

		PAGE
APPENDIX		
1	Dried saba banana peel powder	49
2	Preparation of SBP biosorbent	49
3	Preparation of SBP biosorbent prior to adsorption experiment	50
4	Labeling the beakers	50
5	Preparation of Chromium solution	51
6	Preparation of SBP biosorbent used in adsorption	51
7	Adsorption experiment	52
8	Amount of Chromium ions adsorbed by SBP biosorbent	52
9	Ion uptake capacity	53
10	Percentage of removed Chromium (IV)	53

## ABSTRACT

**ABRACIA, ROSE MARIE C.**, Department of Agricultural and Biosystems Engineering, College of Engineering, Central Luzon State University, Science City of Munoz, Nueva Ecija, Philippines, **June 2023**, **“SABA BANANA (*Musa acuminata*) PEEL AS BIOSORBENT OF CHROMIUM (Cr) HEAVY METAL FROM AQUEOUS SOLUTION”**

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The production of food for humans and animals, as well as industrial and agricultural processes, all depend increasingly on clean water. As a result of rapid industrialization, heavy metals including chromium are in our waterways. The goal of this study was to help with the search for less expensive water treatment methods, that is why Saba Banana Peels are used as a biosorbent. SBP has a rough structure with small openings and holes on its surface which increased the surface area and allowed pores to diffuse during the adsorption process. The adsorption experiment conducted reveals that the amount of chromium ions adsorbed is from 1966.67mg/L to 7966.67mg/L, and as the initial ion concentration rises, so does the amount of adsorbed ions. Also, the uptake capacity of the SBP biosorbent is from 98.33mg/g to 398.17mg/g, and it was observed that uptake capacity increases as the initial concentration of chromium increases. For the percentage of adsorption, it starts from 98.33% and stops at 99.59% after 48 hours of contact time. Which indicates that percentage of Cr (VI) removed from aqueous solution increases when the chromium concentration also increases, it was concluded that SBP is a potential biosorbent for heavy metal like chromium which can be commonly found in wastewaters.

Keywords: biosorbent, saba banana peels, Chromium (VI), heavy metal

# INTRODUCTION

## Background of the study

Earth is covered by water, and it is necessary for all living things to survive. It supports the growth of the economy, the production of food, and human health. Despite the abundance of water, it is still difficult to get access to safe drinking water. Water that is unsafe or contaminated with toxic or dangerous substances should not be consumed by plants or animals because it puts their health at risk. The environment has been under significant stress for the past century, and water pollution has gotten worse as industries have expanded and developed quickly.

Water pollution is defined as the release of contaminants into surface groundwater, lakes, streams, rivers, estuaries, and oceans to the point where they prevent ecosystems from functioning normally or prevent beneficial uses of the water. It can occur because of the discharge of energy into bodies of water as well as the release of substances such as

## LITERATURE CITED

- Aguda, F.A., Mercado, L.E., Santos, P.M., Ondoy, K. & Macale, A. 2021. The Adsorption Efficacy of Activated Carbon from Rice Husks (*Oryza sativa* L.) and Saba Banana (*Musa acuminata* x *balbisiana*) Peelings for the Removal of Chromium from Laboratory Wastewater. *Journal of Nature Studies*. 20(1), 54-67.
- Akpomie, K. G., & Conradie, J. (2020). Banana peel as a biosorbent for the decontamination of water pollutants. A review. *Environmental Chemistry Letters*, 18(4), 1085–1112. <https://doi.org/10.1007/s10311-020-00995-x>
- Aminuzzaman, M., Kei, L. M., & Liang, W. H. (2017). Green synthesis of copper oxide (CuO) nanoparticles using banana peel extract and their photocatalytic activities. *AIP Conference Proceedings*. <https://doi.org/10.1063/1.4979387>
- Anwar, J., Shafique, U., Waheeduz, Z., Salman, M., Dar, A., & Anwar, S. (2010). Removal of Pb(II) and Cd(II) from Water by Adsorption on Peels of Banana. *Bioresource Technology*, 101(6), 1752-1755. doi. org/10.1016/j.biortech.2009.10.021
- Arivoli S., Hema M., and Prasath P. (2009). Adsorption of Malachite Green onto Carbon prepared from Borassus Bark. *Arabian Journal for Science & Engineering* (SpringerScience & Business Media BV), vol. 34, 2009.
- Ashraf, M. A., Maah, M. J., & Yusoff, I. (2010). Study of mango biomass (*Mangifera indica* L) as a cationic biosorbent. *International Journal of Environmental Science & Technology*, 7(3), 581–590. doi:10.1007/bf03326167
- Boniolo, M.R. (2008). Biossorção de Urânio nas cascas de banana. Dissertação de Mestrado – Instituto de Pesquisas Energéticas e Nucleares, São Paulo – Spain
- Brochin R, Leone S, Phillips D, Shepard N, Zisa D, Angerio A. The cellular effect of lead poisoning and its clinical picture. *GUJHS*. 2008;5(2):1–8.
- Chakraborty, R. Asthana, A. Singh, A. (2020). Adsorption of heavy metal ions by various low-cost adsorbents: a review. <https://doi.org/10.1080/03067319.2020.1722811>
- Goswami, R., Shim, J., Deka, S., Kumari, D., Kataki, R., & Kumar, M. (2016). Characterization of cadmium removal from aqueous solution by biochar produced from *Ipomoea fistulosa* at different pyrolytic temperature. *Ecological Engineering*, 97, 444-451.
- Gunatilake, S.K. (2015). Methods of removing heavy metals from industrial water. *Journal of Multidisciplinary Engineering Science Studies*, 1(1), 12-18.

- Gupta, A., Verma, S., Khan, K. and Verma, R. (2013). Phytoremediation Using Aromatic Plants: A Sustainable Approach for Remediation of Heavy Metals Polluted Sites. *Environ. Sci. Technol.* 2013, 47, 18, 10115–10116 <https://doi.org/10.1021/es403469c>
- Hima, K., Srinivasan, S., Vijaya, S., Jayakumar, S., Suryanarayana, V., Venkateshwar, P. (2007). Biosorption: An ecofriendly alternative for heavy metal removal. *African Journal of Biotechnology* 6 (25); 2924-2931.
- Housagul, S., Sirisukpoka, U., Boonyawanich, S., Pisutpaisal, N., 2014. Biomethane production from Co-digestion of banana peel and waste glycerol. *Energy Proced*, 61, 2219-2223. <https://doi.org/10.1016/j.egypro.2014.12.113>.
- Ibrahim, U. K., Kamarrudin, N., Suzihaque, M. U. H., & Abd Hashib, S. (2017). Local Fruit Wastes as a Potential Source of Natural Antioxidant: An Overview. *IOP Conference Series: Materials Science and Engineering*, 206, 012040. doi:10.1088/1757-899x/206/1/012040
- Inyang, M., Gao, B., Yao, Y., Xue, Y., Zimmerman, A.R., Pullammanappallil, P., et al. (2012). Removal of heavy metals from aqueous solution by biochars derived from anaerobically digested biomass. *Bioresour Technol*, 110, 50-56.
- Jaishankar, M., Tseten, T., Anbalagan, N., Mathew, B. B., & Beeregowda, K. N. (2014). Toxicity, mechanism, and health effects of some heavy metals. *Interdisciplinary toxicology*, 7(2), 60–72. <https://doi.org/10.2478/intox-2014-0009>
- Javid, A., Bajwa, R. and Manzoor, T. 2011. Biosorption of heavy metals by pretreated biomass of *Aspergillus niger*. *Pakistan Journal of Botany*. 43 (1): 419-425
- Koutras GA, Schneider AS, Hattori M, Valentine WN. Studies on chromated erythrocytes. Mechanisms of chromate inhibition of glutathione reductase. *Br J Haematol*. 1965;11(3):360–369.
- Lakherwal. D. (2014). Adsorption of heavy metals: a review. *International journal of environment research and development*., 4(1), 41-48.
- Lam, S.S., Liew, R. K., Cheng, C.K., Rasit, N., C. K. Ooi., N. L. Ma., Ng, J. H., Lam, W. H., Chong, C.T., Chase, H. A., 2018. Pyrolysis production of fruit peel biochar for potential use in treatment of palm oil mill effluent. *J. Environ. Manag.* 213, 400-408. <https://doi.org/10.1016/j.jenvman.2018.02.092>.

- Lingamdinne, L., Koduru, J., Jyothi, R., Chang, Y., & Yang, J. (2015). Factors effect on bioremediation of Co(II) and Pb(II) onto *Lonicera japonica* flowers powder. *Desalination Water Treatment*. doi: 10.1080/19443994.2015.1055813
- Martin S, Griswold W. Human health effects of heavy metals. *Environmental Science and Technology Briefs for Citizens*. 2009;(15):1–6.
- Markowitz M. Lead Poisoning. *Pediatr Rev*. 2000;21(10):327–335.
- Matsumoto ST, Mantovani MS, Malaguttii MIA, Dias AL, Fonseca IC, Marin-Morales MA. Genotoxicity and mutagenicity of water contaminated with tannery effluents, as evaluated by the micronucleus test and comet assay using the fish *Oreochromis niloticus* and chromosome aberrations in onion root-tips. *Genet Mol Biol*. 2006;29(1):148–158.
- Momcilovic M., Purenovic M., Bojic A., Zarubica A., Randelovic, M. (2011). Removal of Lead (II) ions from aqueous solutions by adsorption onto pinecone activated carbon. *Desakin*, 276(1), 53-59.
- Monachese, M., Burton, J., & Reida, G. (2012). Bioremediation and tolerance of humans to heavy metals through microbial processes: A potential role for probiotics. *Appl. Environ. Microbiol.* (78), 6397-6404.
- Najeeb U, Ahmad W, Zia MH, Malik Z, Zhou W. (2014). Enhancing the lead phytostabilization in wetland plant *Juncus effusus* L. through somaclonal manipulation and EDTA enrichment. *Arab J Chem*.
- Naggarajaiah, S.B. and Aprakash, J. (2011). Chemical composition and antioxidant potential of peels from three varieties of banana. *Asian journal of food and agro-industry*, 4 (01), 31-46
- Okino-Delgado, C. H., Fleuri, L. F., 2015. Orange and mango byproducts: agro-industrial waste as source of bioactive compounds and botanical versus commercial description—A review. *Food Reviews Intern*. 32, 1–14. <https://doi.org/10.1080/87559129.2015.1041183>.
- O'Brien T, Xu J, Patierno SR. Effects of glutathione on chromium-induced DNA crosslinking and DNA polymerase arrest; *Molecular Mechanisms of Metal Toxicity and Carcinogenesis*; US: Springer; 2001. pp. 173–182.
- O'Connell, D., Birkinshaw, C. & O'Dwyer, T. (2008). Heavy metal adsorbents prepared from the modification of cellulose: A review, *Bioresource Technology*, Volume 99,

Issue 15, Pages 6709-6724, ISSN 0960-8524,  
<https://doi.org/10.1016/j.biortech.2008.01.036>.

- Papanikolaou NC, Hatzidaki EG, Belivanis S, Tzanakakis GN, Tsatsakis AM. Lead toxicity update. A brief review. *Med Sci Monitor*. 2005;11(10):RA329.
- Parfitt, J., Barthel, M., Macnaughton, S., Godfray, H.C.J., Beddington, J.R., Crute, I.R., Haddad, L., Lawrence, D., Muir, J.F., Pretty, J., 2010. Food waste within food supply chains: quantification and potential for change to 2050. *Phil. Trans. R. Soc. B*. 365, 3065-3081. <https://doi.org/10.1098/rstb.2010.0126>.
- Pokharel, P., Kwak, J. H. Ok, Y.S., Chang, S. X., 2018. Pine sawdust biochar reduces GHG emission by decreasing microbial and enzyme activities in forest and grassland soils in a laboratory experiment. *Sci. Total Environ*. 625, 1247-1256. <https://doi.org/10.1016/j.scitotenv.2017.12.343>.
- Prusty, P.K. & Satapathy, K.B. (2020). Phytoremediation of Wastewater by Using *Azolla anabaena* Consortium and its Aquatic Associates: A Review. *Plant Archives*, 20(1), 1933-1943. Retrieved July 13, 2022, from [http://www.plantarchives.org/20-1/1933-1943%20\(5990\).pdf](http://www.plantarchives.org/20-1/1933-1943%20(5990).pdf)
- Reginio, F. C., Qin, W., Ketnawa, S., & Ogawa, Y. (2020). Bio-properties of Saba banana (*Musa 'saba'*, ABB Group): Influence of maturity and changes during simulated in vitro gastrointestinal digestion. *Scientific Reports*, 10(1). <https://doi.org/10.1038/s41598-020-63501-x>
- Renu; Madhu Agarwal; K. Singh. (2017). Heavy metal removal from wastewater using various adsorbents: a review. <https://doi.org/10.2166/wrd.2016.104>
- Sial, T. A., Khan, M. N., Lan, Z., Kumbhar, F., Zhao, Y., Zhang, J., Xiu, L. (2018). *Contrasting effects of banana peels waste and its biochar on greenhouse gas emissions and soil biochemical properties*. *Process Safety and Environmental Protection*. doi:10.1016/j.psep.2018.10.030
- Thürmer K, Williams E, Reutt-Robey J. Autocatalytic oxidation of lead crystallite surfaces. *Science*. 2002;297(5589):2033–2035.
- Tripathi A, Ranjan MR (2015) Heavy Metal Removal from Wastewater Using Low-Cost Adsorbents. *J Bioremed Biodeg* 6:315. doi:10.4172/2155-6199.1000315
- United States Environmental Protection Agency. (2016). Retrieved July 13, 2022, from <https://www.epa.gov/superfund/lead-superfund-sites-human-health>

- Wang, J., & Chen, C. (2009). Bio sorbents for heavy metals removal and their future. *Biotechnology Advances* (27), 195-226.
- Wołowiec, M., Komorowska-Kaufman, M., Pruss, A., Rzepa, G., & Bajda, T. (2019). Removal of Heavy Metals and Metalloids from Water Using Drinking Water Treatment Residuals as Adsorbents: A Review. *Minerals*, 9(8), 487. <https://doi.org/10.3390/min9080487>
- Wu, Y., Ma, H., Zheng, M., Wang, K., 2015. Lactic acid production from acidogenic fermentation of fruit and vegetable wastes. *Biores. Techno.* 191, 53. <https://doi.org/10.1016/j.biortech.2015.04.100>.
- Yongsheng W, Qihui L, Qian T. Effect of Pb on growth, accumulation, and quality component of tea plant. *Procedia Engineering*. 2011;18:214–219.
- Zhao, L., Wang, J., Zhang, P., Gu, Q., & Gao, C. (2018). Absorption of Heavy Metal Ions by Alginate. *Bioactive Seaweeds for Food Applications*, 255–268. doi:10.1016/b978-0-12-813312-5.00013-3
- Zhitkovich A. Importance of chromium-DNA adducts in mutagenicity and toxicity of chromium (VI) *Chem Res Toxicol*. 2005;18(1):3–11.